

Acylation of *o*-Imidoylphenol Lithium Dianions: Synthesis of 4H-Chromen-4-ylidenamines

Cristina Cimarelli and Gianni Palmieri*

Dipartimento di Scienze Chimiche, Università di Camerino, Via S. Agostino 1, I-62032 Camerino, Italy Received 4 October 1999; revised 8 November 1999; accepted 18 November 1999

Abstract—A method is described to obtain 4*H*-chromen-4-ylidenamines **4**, through the reaction of *o*-imidoyl phenol dianions 2' with aromatic esters and subsequent acid cyclisation of the 3-(2-hydroxyphenyl)-3-(amino)-1-phenylprop-2-en-1-ones **3** obtained. © 2000 Elsevier Science Ltd. All rights reserved.

Chromones and flavones belong to a widely explored family of organic compounds largely present in the plant kingdom. Flavones and flavonoids show a general tendency to have biological activity and many compounds have found uses in medicine and industry.¹ 4*H*-Chromen-4-ylidenamines **4** are derivatives of flavones, almost absent from the literature, that have awakened new interest because the simplest of them (2-phenyl-4*H*-chromen-4-imine), has been the object of patents² for treatment of cell proliferative diseases and for its antihypoxic, hypotensive, and antiallergic properties.

We now report the use of o-imidoylphenols **2** as precursors of 4H-chromen-4-ylideneamines **4**, through the reaction of their dianions with esters, as described in Scheme 1.

Several papers show that it is not possible to prepare 4H-chromen-4-ylidenamines **4** by condensation of the corresponding chromones with the relevant amine.³ The reaction of chromone **6** with ammonia and primary and

secondary amines results in the opening of the pyran ring, yielding the corresponding enamino ketone 7, which reverts the starting material in acidic medium as shown in Scheme 2(a). On the other hand Zagorevskii et al. prepared iminochromenes 9 in a very low yield by treatment of 4,4-dichlorochromene 8 with primary amines as in Scheme 2(b).⁴

o-Imidoylphenols **2** can be readily prepared by condensation⁵ of the appropriate *o*-acyl phenol **1** and the amine according to a described procedure.⁶ Upon treatment with LDA (lithium diisopropylamide) they give O,C-dianions 2'that can react with electrophiles. We have found that alkylated *o*-imidoylphenols can be obtained by alkylation of these dianions with alkyl halides.⁷ In the present case the dianion is allowed to react with aromatic esters and in all the reactions we have isolated the acylation products 3-(2-hydroxyphenyl)-3-(amino)-1-phenylprop-2-en-1-ones **3**. We have found the best results in the acylation reaction with an excess of 3.5 mol of LDA per mole of starting



Scheme 1. (i) RNH₂, benzene, H⁺, reflux; (ii) LDA, 0°C, 1 h; (iii) ArCOOR', -80°C, 2 h; (iv) H₃O⁺, CH₂Cl₂, rt; (v) H₂O/THF/AcOH=2/2/1, 40°C, 0.5 h.

Keywords: imidoylphenol; chromenes; acylation; cyclisation.

^{*} Corresponding author. Tel.: +39-737-402241; fax: +39-737-637345; e-mail: palmieri@camserv.unicam.it



Scheme 2.

material: 2 equiv. are necessary for proton abstraction in forming the dianion 2', while another metallates the product 3' preventing proton abstraction by 2' more basic that 3''.

The yields obtained in the acylation step (Table 1) are not very high and moreover are not constant for the same reaction in different preparations: this probably happens because during the chromatographic purification a partial cyclisation take place, that we have not quantified. However, cyclisation by aqueous AcOH, before chromatographic purification of the crude reaction mixture containing the acylation product **3ca**, resulted in a 78% of yield of the final 4*H*-chromen-4-ylidene amine **4ca**.

Cyclisation also takes place on heating the acylated product. Enaminoketone **3aa** in refluxing toluene after 0.5 h produces the chromenimine **4aa** (64% yield) with a partial decomposition. For this reason we were unable to register the mass spectra of all acylated products **3** because GC-MS analysis showed for every product **3** the mass spectrum of the corresponding chromen imine **4**.

Attempts to perform the reaction with aliphatic esters were unsuccessful. In the reactions with ethyl propionate and ethyl phenylacetate, although we could observe in the mass spectra of both crude reaction mixtures the peaks corresponding to the chromen imines, all attempts to isolate the products failed, as the mixtures decomposed after a few minutes to a blue oil not containing any relevant product.

The purified 3-(2-hydroxyphenyl)-3-(amino)-1-phenylprop-2-en-1-ones**3**were heated in acid solution and in all cases a single product was isolated in almost quantitative yield

Table 1. Acylation of *o*-imidoyl phenol **2** and cyclisation of the intermediate hydroxyphenyl enaminone **3** to 4H-chromen-4-ylidene amines **4**

2	R	Ar	3	Yield ^a (%)	4	Yield ^a (%)
2a	Me	Ph	3aa	72	4aa	92 92
2a 2a	Me Me	4-MeO-Ph 4-Pv	3ab 3ac	63 67	4ab 4ac	85 66
2b	Pr ⁱ	Ph	3ba	42	4ba	72
2b 2c	Pr' Bn	4-MeO-Ph Ph	3bb 3ca	47 64	4bb 4ca	81 94 (78) ^b
2c	Bn	4-MeO-Ph	3cb	37	4cb	87
2c	Bn	4-Py	3cc	36	4cc	76

^a Of the pure isolated product.

^b Yield of the one pot preparation starting from the crude reaction mixture of **3ca**.

(Table 1) that was shown by spectroscopy to have a *4H*-chromen-4-ylidenamine structure **4**.

In summary, starting from cheap and commercially available starting materials, with known easily performed procedures, we have found a useful strategy for the preparation of 4H-chromen-4-ylidenamines **4** in satisfactory yields, a class of compounds of renewed interest.

Experimental

¹H and ¹³C NMR spectra were recorded with a Varian VXR 300 instrument. Chemical shifts are given in ppm downfield from Me₄Si in CDCl₃ solution. Coupling constants are given in Hertz. IR spectra were recorded with a Perkin–Elmer 257 spectrometer. GC-MS analyses were performed with an HP 59970 workstation formed by an HP-5890 gas chromatograph equipped with a methyl silicone capillary column and by an HP-5970 mass detector. All melting points are uncorrected. THF was dried by refluxing over sodium wires until the blue colour of benzophenone ketyl persisted and then distilled into a dry receiver under nitrogen atmosphere. All reagents and solvents were distilled prior to use or were of commercial quality from freshly opened containers. Commercial butyllithium solutions (Aldrich) were employed under dry atmosphere.

Preparation of starting 2-imidoyl phenols 2a-c

The 2-imidoyl phenols $2\mathbf{a} - \mathbf{c}$ were prepared by condensation of the appropriate *o*-acylphenol⁴ and the amine according to described procedure.⁵

Acylation of dianions of 2-imidoyl phenols 2a-c

2-Imidoyl phenols 2a-c (3 mmol) were dissolved in anhydrous THF (5 cm³) under nitrogen atmosphere at 0°C and treated with a solution in THF of LDA, obtained by mixing butyllithium (10.5 mmol) with an equimolecular amount of diisopropylamine in THF (3 cm³). The reaction mixture was allowed to stand for 1 h, then was cooled to -80° C and treated with a solution of an aromatic ester (10.5 mmol) in THF (2 cm³). The reaction was monitored by TLC with *n*-hexane/ethyl acetate=70/30 as eluent and after a variable time (2-3 h), when the starting material was consumed, was quenched with NH₄Cl saturated solution (5 cm³) and brine (2 cm³) and extracted with CH₂Cl₂ $(2\times 20 \text{ cm}^3)$. The solution was dried with anhydrous Na₂SO₄ then filtered and the solvent evaporated under reduced pressure. Chromatographic separation of the crude oil obtained, with cyclohexane/ethyl acetate=70/30 as eluent afforded the acylated product **3** in 36–72% yield.

(Z)-3-(2-Hydroxyphenyl)-3-(methylamino)-1-phenylprop-2-en-1-one (3aa). Yellow crystals; mp 128–132°C (CH₂Cl₂/ *n*-hexane); ν_{max} (Nujol) 1608, 1579, 1342, 1259, 769, 686 cm⁻¹. NMR (CDCl₃): $\delta_{\rm H}$ 2.85 (d, 3 H, *J*=5.3 Hz), 5.78 (s, 1 H), 6.50 (br s, 1 H), 6.92–7.86 (m, 9 H), 11.27 (br s, 1 H). $\delta_{\rm C}$ 31.27, 93.44, 116.84, 122.28, 126.00, 127.50, 128.00, 129.00, 130.50, 131.29, 140.31, 153.36, 154.17, 188.64. Anal. Calcd for C₁₆H₁₅NO₂, MW 253.296: C, 75.87; H, 5.97; N, 5.53%. Found: C, 75.76; H, 5.80; N, 5.68%.

(Z)-3-(2-Hydroxyphenyl)-1-(4-methoxyphenyl)-3-(methylamino)prop-2-en-1-one (3ab). Oil; ν_{max} (liquid film) 1909, 1709, 1367, 878 cm⁻¹. NMR (CDCl₃): $\delta_{\rm H}$ 2.86 (d, 3 H, J=5.3 Hz), 3.85 (s, 3 H), 5.80 (s, 1 H), 6.40 (br s, 1 H), 6.83–7.83 (m, 8 H), 11.10 (br s, 1 H). $\delta_{\rm C}$ 31.11, 55.36, 92.87, 113.47, 117.09, 120.71, 122.42, 129.31, 129.48, 131.58, 133.07, 153.72, 162.41, 163.87, 188.56. Anal. Calcd for C₁₇H₁₇NO₃, MW 283.322: C, 72.07; H, 6.05; N, 4.94%. Found: C, 72.20; H, 6.12; N, 4.68%.

(Z)-3-(2-Hydroxyphenyl)-3-(methylamino)-1-pyridin-4ylprop-2-en-1-one (3ac). Orange oil; ν_{max} (liquid film) 1600, 1580, 1240, 754, 681 cm⁻¹. NMR (CDCl₃): $\delta_{\rm H}$ 2.97 (d, 3 H, *J*=5.3 Hz), 3.41 (br s, 1 H), 5.72 (s, 1 H), 6.82–8.85 (m, 8 H), 11.60 (br d, 1 H, *J*=5.2 Hz). $\delta_{\rm C}$ 31.74, 93.58, 117.09, 118.38, 120.73, 129.42, 131.90, 147.62, 150.33, 151.23, 153.88, 166.90, 185.70. Anal. Calcd for C₁₅H₁₄N₂O₂, MW 254.284: C, 70.85; H, 5.55; N, 11.02%. Found: C, 70.92; H, 5.87; N, 10.89%.

(Z)-3-(2-Hydroxyphenyl)-3-(isopropylamino)-1-phenylprop-2-en-1-one (3ba). Yellow crystals, mp 136–138°C (CH₂Cl₂/*n*-hexane); ν_{max} (Nujol) 2715, 1719, 1382, 1121, 1074, 813, 712 cm⁻¹. NMR (CDCl₃): $\delta_{\rm H}$ 1.33 (d, 6 H, J=6.3 Hz), 3.60 (d hept, 1 H, J=6.3, 9.6 Hz), 5.77 (s, 1 H), 6.60–7.89 (m, 9 H), 11.20 (d, 1 H, J=9.6 Hz). $\delta_{\rm C}$ 24.04, 47.41, 94.10, 117.22, 121.22, 123.75, 128.17, 129.37, 130.06, 132.00, 133.89, 141.38, 154.50, 162.10, 188.30. Anal. Calcd for C₁₈H₁₉NO₂, MW 281.349: C, 76.84; H, 6.81; N, 4.98%. Found: C, 76.57; H, 6.53; N, 5.14%.

(Z)-3-(2-Hydroxyphenyl)-3-(isopropylamino)-1-(4-methoxyphenyl)prop-2-en-1-one (3bb). Oil; ν_{max} (liquid film) 1734, 1567, 1458, 1022, 973, 850, 797 cm⁻¹. NMR (CDCl₃): $\delta_{\rm H}$ 1.19 (d, 6 H, *J*=6.2 Hz), 3.53 (d hept, 1 H, *J*=6.3, 9.8 Hz), 3.84 (s, 3 H), 5.75 (s, 1 H), 6.60 (br s, 1 H), 6.86–7.05 (m, 4 H), 7.20–7.40 (m, 2 H), 7.80–7.90 (m, 2 H), 11.02 (br d, 1 H, *J*=9.8 Hz). $\delta_{\rm C}$ 23.87, 46.73, 55.35, 93.32, 113.47, 116.30, 120.33, 122.40, 128.56, 129.04, 130.95, 132.48, 152.91, 161.02, 162.01, 188.25. Anal. Calcd for C₁₉H₂₁NO₃, MW 311.375: C, 73.29; H, 6.80; N, 4.50%. Found: C, 73.54; H, 6.65; N, 4.74%.

(Z)-3-(Benzylamino)-3-(2-hydroxyphenyl)-1-phenylprop-2-en-1-one (3ca). Yellow crystals, mp 174–176°C (ethyl acetate/*n*-hexane). ν_{max} (Nujol) 1605, 1560, 1352, 1309, 765, 691 cm⁻¹. NMR (CDCl₃): $\delta_{\rm H}$ 4.35 (d, 2 H, *J*=6.2 Hz), 5.87 (s, 1 H), 6.40 (br s, 1 H), 6.90–7.90 (m, 14 H), 11.55 (br s, 1 H). $\delta_{\rm C}$ 49.03, 94.45, 117.02, 121.01, 122.16, 125.16, 126.05, 127.68, 128.03, 128.77, 129.24, 129.35, 131.63, 131.78, 136.22, 137.13, 153.35, 163.00. Anal. Calcd for C₂₂H₁₉NO₂, MW 329.392: C, 80.22; H, 5.81; N, 4.25%. Found: C, 79.97; H, 6.03; N, 4.40%.

(Z)-3-(Benzylamino)-3-(2-hydroxyphenyl)-1-(4-methoxyphenyl)prop-2-en-1-one (3cb). Yellow crystals, mp 154°C (CH₂Cl₂/*n*-hexane); ν_{max} (Nujol) 1733.8, 1611.7, 1565.9, 1247.7, 1168.0, 699.2 cm⁻¹. NMR (CDCl₃): $\delta_{\rm H}$ 3.83 (s, 3 H), 4.31 (d, 2 H, *J*=6.3 Hz), 5.83 (s, 1 H), 6.60 (br s, 1 H), 6.80–7.90 (m, 13 H), 11.50 (br s, 1 H). $\delta_{\rm C}$ 48.66, 55.54, 93.96, 113.65, 116.78, 120.46, 122.00, 127.40, 127.64, 128.89, 129.156, 129.37, 131.35, 132.58, 138.26., 153.40, 162.25, 162.61, 188.62. Anal. Calcd for C₂₃H₂₁NO₃, MW 359.418: C, 76.86; H, 5.89; N, 3.90%. Found: C, 77.03; H, 5.74; N, 3.84%.

(Z)-3-(Benzylamino)-3-(2-hydroxyphenyl)-1-pyridin-4-ylprop-2-en-1-one (3cc). Oil; ν_{max} (liquid film) 1596, 1566, 1537, 1413, 1341, 1313, 1148, 1007, 909 cm⁻¹. NMR (CDCl₃): $\delta_{\rm H}$ 4.46 (s, 2 H), 5.78 (s, 1 H), 6.50 (br s, 1 H), 6.80–8.70 (m, 13 H), 11.97 (br s, 1 H). $\delta_{\rm C}$ 48.88, 93.71, 116.31, 119.83, 119.95, 121.40, 122.07, 127.46, 127.63, 128.74, 129.29, 131.40, 137.44, 147.83, 149.18, 154.25, 166.58, 184.76. Anal. Calcd for C₂₁H₁₈N₂O₂, MW 330.380: C, 76.34; H, 5.49; N, 8.48%. Found: C, 76.61; H, 5.77; N, 8.36%.

Cyclisation of 3-(2-hydroxyphenyl)-3-(amino)-1-phenylprop-2-en-1-ones 3

The enaminoketones **3** obtained from the previous step (3 mmol) were dissolved in a water/THF/acetic acid solution in the ratio 2/2/1 (10 cm³) and heated at 40°C for 0.5 h. After this time the mixture was neutralised with saturated solution of Na₂CO₃ and extracted with CH₂Cl₂ (2×30 cm³). The solution was dried with anhydrous Na₂SO₄, then filtered and the solvent evaporated under reduced pressure. The products obtained were submitted to chromatographic purification or recrystallised. In all cases a single product was isolated, in 66–94% yield, which spectral data identified as a 4*H*-chromen-4-ylidenamine **4**. Prolonged heating of the mixture (5 h) results only in a little decomposition of the product to unidentified products.

N-Methyl-*N*-(2-phenyl-4*H*-chromen-4-ylidene)amine (4aa). Yellow oil; ν_{max} (liquid film) 2854, 1636, 1221, 984, 762, 689 cm⁻¹. NMR (CDCl₃): $\delta_{\rm H}$ 3.38 (s, 3 H), 6.75 (s, 1 H), 7.20–7.40 (m, 2 H), 7.42–7.63 (m, 4 H), 7.80–8.00 (m, 2 H), 8.18–8.22 (m, 1 H). $\delta_{\rm C}$ 38.15, 95.54, 118.02, 124.57, 125.27, 126.33, 129.25, 130.98, 131.43, 133.57, 134.50, 153.50, 154.00, 157.50; *m*/*z* 235 (M⁺, 68), 234 (100); 206 (13); 105 (12); 77 (13). Anal. Calcd for C₁₆H₁₃NO, MW 235.281: C, 81.68; H, 5.57; N, 5.95%. Found: C, 81.97; H, 5.76; N, 5.87%.

N-[2-(4-Methoxyphenyl)-4*H*-chromen-4-ylidene]-*N*methylamine (4ab). Yellow oil; ν_{max} (liquid film) 2896, 1640, 1220, 769, 702 cm⁻¹. NMR (CDCl₃): $\delta_{\rm H}$ 3.36 (s, 3 H), 3.87 (s, 3 H), 6.65 (s, 1 H), 6.96–7.01 (m, 2 H), 7.20–7.50 (m, 3 H), 7.80–7.90 (m, 2 H), 8.20 (d, 1 H, J=9.4 Hz). $\delta_{\rm C}$ 37.88, 55.92, 95.17, 114.66, 117.92, 123.44, 124.63, 125.20, 125.89, 127.92, 131.39, 153.54, 154.14, 157.46, 162.03. m/z 265 (M⁺, 55), 264 (72), 221 (25), 112 (36), 105 (32), 84 (100). Anal. Calcd for C₁₇H₁₅NO₂, MW 265.307: C, 76.96; H, 5.70; N, 5.28%. Found: C, 76.78; H, 5.87; N, 5.44%.

N-Methyl-*N*-(2-pyridin-4-yl-4*H*-chromen-4-ylidene)amine (4ac). Red oil; ν_{max} (liquid film) 2898, 1621, 1120, 972, 753, 685 cm⁻¹. NMR (CDCl₃): $\delta_{\rm H}$ 3.35 (s, 3 H), 6.81 (s, 1 H), 7.20–8.75 (m, 8 H). $\delta_{\rm C}$ 30.00, 98.12, 115.55, 119.17, 121.12, 128.33, 128.94, 137.05, 138.22, 152.44, 154.09, 158.64, 164.18. Anal. Calcd for C₁₅H₁₂N₂O, MW 236.269: C, 76.25; H, 5.12; N, 11.86%. Found: C, 76.32; H, 5.27; N, 11.69%.

N-(2-Phenyl-4*H*-chromen-4-ylidene)propan-2-amine (4ba). Yellow oil; ν_{max} (liquid film) 2901 1611, 1498, 1231, 863, 691 cm⁻¹. NMR (CDCl₃): $\delta_{\rm H}$ 1.31 (d, 6 H, *J*=6.3 Hz), 4.00 (hept, 1 H, *J*=6.3 Hz), 6.75 (s, 1 H), 7.26–7.85 (m, 9 H). $\delta_{\rm C}$ 23.71, 49.29, 96.38, 117.38, 121.58, 124.31, 124.84, 125.88, 128.79, 130.54, 131.13, 132.28, 146.12, 151.24, 157.06. Anal. Calcd for C₁₈H₁₇NO, MW 263.334: C, 82.10; H, 6.51; N, 5.32%. Found: C, 82.36; H, 6.74; N, 5.23%.

N-[2-(4-Methoxyphenyl)-4*H*-chromen-4-ylidene]propan-2-amine (4bb). Yellow crystals, mp 86–88°C (CH₂Cl₂/*n*hexane); ν_{max} (Nujol) 1968, 1608, 1510, 1218, 1116, 830, 790, 664 cm⁻¹. NMR (CDCl₃): $\delta_{\rm H}$ 1.30 (d, 6 H, *J*=3.1 Hz), 3.88 (s, 3 H), 3.70–4.10 (m, 1 H), 6.65 (s, 1 H), 6.95–8.40 (m, 8 H). $\delta_{\rm C}$ 24.30, 50.00, 56.30, 96.00, 115.37, 117.42, 118.49, 124.19, 127.20, 127.58, 128.73, 146.03, 151.04, 154.00, 156.80, 162.15. Anal. Calcd for C₁₉H₁₉NO₂, MW 293.360: C, 77.79; H, 6.53; N, 4.77%. Found: C, 77.91; H, 6.70; N, 4.54%.

N-Benzyl-*N*-(2-phenyl-4*H*-chromen-4-ylidene)amine (4ca). Yellow crystals, mp 94–96°C (CH₂Cl₂/*n*-hexane). ν_{max} (Nujol) 1633, 1575, 1412, 1349, 1072, 821, 760 cm⁻¹. NMR (CDCl₃): $\delta_{\rm H}$ 4.83 (s, 2 H), 6.82 (s, 1 H), 7.20–8.50 (m, 14 H). $\delta_{\rm C}$ 53.98, 96.50, 117.50, 124.71, 124.82, 125.59, 125.92, 126.50, 127.13, 127.65, 128.40, 128.80, 130.59, 131.14, 133.06, 141.00, 152.63, 153.26. Anal. Calcd for C₁₉H₁₉NO₂, MW 311.377: C, 84.86; H, 5.50; N, 4.50%. Found: C, 84.92; H, 5.32; N, 4.46%.

N-Benzyl-*N*-[2-(4-methoxyphenyl)-4*H*-chromen-4-ylidene]amine (4cb). Yellow crystals, mp 132–133°C (CH₂Cl₂/*n*hexane); ν_{max} (Nujol) 1630, 1608, 1572, 1288, 1179, 1077, 983, 958, 917, 902, 874, 855, 818 cm⁻¹. NMR (CDCl₃): $\delta_{\rm H}$ 3.88 (s, 3 H), 4.82 (s, 2 H), 6.72 (s, 1 H), 7.00 (d, 2 H, J=8.8 Hz), 7.30–7.41 (m, 5 H), 7.47–7.53 (m, 3 H), 7.82 (d, 2 H, J=8.9 Hz), 8.42 (d, 1 H, J=8.1 Hz). $\delta_{\rm C}$ 53.67, 55.46, 95.33, 114.18, 117.38, 117.78, 124.70, 125.37, 126.47, 127.49, 127.54, 127.67, 128.40, 131.03, 133.18, 141.13, 152.68, 153.24, 161.57. Anal. Calcd for C₂₃H₁₉NO₂, MW 341.402: C, 80.92; H, 5.61; N, 4.10%. Found: C, 80.82; H, 5.87; N, 4.34%.

4*H***-Benzyl***-N***-(2-pyridin-4-yl-4***H***-chromen-4-ylidene)methanamine (4cc).** Orange crystals, mp 102–104°C (CH₂Cl₂/*n*-hexane); ν_{max} (Nujol) 1611, 1573, 1546, 1495, 1325, 1229, 863, 839, 679, 661 cm⁻¹. NMR (CDCl₃): $\delta_{\rm H}$ 4.82 (s, 2 H), 6.91 (s, 1 H), 7.26–8.75 (m, 13 H). $\delta_{\rm C}$ 53.89, 98.37, 117.46, 119.56, 123.61, 124.76, 125.24, 126.66, 127.60, 128.49, 131.46, 140.31, 141.23, 150.54, 152.03, 153.12, 154.52. Anal. Calcd for C₂₁H₁₆N₂O, MW 312.365: C, 80.75; H, 5.16; N, 8.97%. Found: C, 80.52; H, 5.39; N, 8.84%.

Acknowledgements

Financial support from the Ministero dell'Università e della Ricerca Scientifica e Tecnologica, Rome, and the University of Camerino (National Project "Stereoselezione in Sintesi Organica Metodologie ed applicazioni") are gratefully aknowledged.

References

1. Katritzky, A. R.; Rees, C. W. Comprehensive Heterocyclic Chemistry; Pergamon: Oxford, 1984; p 574.

2. (a) *Chem. Abstr.* 123, 276072 U.S.S.R. SU 1746675 A1, 30 September 1994. Oganesyan, E. T.; Ivchenko, A. V.; Ivashev, M. N.; Lysenko, T. A.; Saraf, A. S. *Izobreteniya* **1994**, *18*, 213. (b) Aggarwal, Bharat B. *Chem. Abstr.* 123, 188575 PCT Int. Appl. WO 9518606 A1, 13 July 1995.

 (a) Wittig, G.; Blumenthal, H. Ber. 1927, 60, 1085. (b) Kostka,
K. Rocz. Chem. 1973, 47, 841. (c) Zagorevskii, V. A.; Orlova, E.
K.; Tsvetkova, I. D. Chem. Heterocycl. Compd. Engl. Transl. 1972, 8, 416. (d) Lockart, I. M.; Tanner, E. M. J. Chem. Soc. 1965, 3610.
Zagorevskii, V. A.; Tsvetkova, I. D.; Orlova, E. K. Chem. Heterocycl. Compd. Engl. Transl. 1967, 8, 624.

- 5. (a) Luche, J. L. J. Am. Chem. Soc. **1978**, 100, 2226. (b) Gemal,
- A. L.; Luche, J. L. J. Am. Chem. Soc. **1970**, 100, 2220. (b) (
- 6. (a) Brown, H. C.; Krishnamurthi, S. Tetrahedron 1979, 35, 567.
- (b) Pearson, R. G. J. Am. Chem. Soc. 1963, 85, 3533.
- 7. (a) Cimarelli, C.; Palmieri, G. Tetrahedron 1998, 54 (51),
- 15 711. (b) Palmieri, G. Eur. J. Org. Chem. 1999, 805.